Richard R. Whitney (1) and David A. Jaeger*

Department of Chemistry, University of Wyoming, Laramie, Wyoming 82071

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Five tetradeuterated analogues (1a, 1b, 1c, 1d, and 1e) of 3,6,9,12,15-pentaoxa-21-aza-bicyclo[15.3.1]heneicosa-1(21),17,19-triene (pyridyl-18-crown-6, 1) have been prepared corresponding to the five unique protonated carbon atoms of the polyether cycle. Deuterium labeling was $\geq 99\%$ for four of the compounds and $\geq 97\%$ for the other (1a).

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Macrocyclic polyethers (crown ethers) have been the subject of numerous reports (2) describing their synthesis and physical and chemical properties. In a study (3) of pyridyl-substituted crown ethers, we required the use of deuterium labeled analogues of 3,6,9,12,15-pentaoxa-21-azabicyclo[15.3.1]heneicosa-1(21),17,19-triene (pyridyl-18-crown-6, 1) (4). We present here the synthesis of five

such analogues. The polyether carbons of 1 are designated a, b, c, d, and e, and the corresponding tetradeuterated compounds 1a, 1b, 1c, 1d, and 1e were prepared. The synthetic procedures employed here should be useful in labeling other crown ether series.

Scheme I

Scheme II

The synthesis of 1a is outlined in Scheme I. Reduction of dimethyl 2,6-pyridinedicarboxylate with sodium borodeuteride in ethanol-O-d yielded 2,6-bis(hydroxymethyl-

 d_2)pyridine (2a). This diol was converted to its disodium salt with sodium hydride in tetrahydrofuran (THF), and reaction of the salt with 3,6,9-trioxaundecane-1,11-diol di-p-toluenesulfonate (3a) yielded 1a. By mass spectrometry 1a contained ≥ 3.88 atoms of excess deuterium per molecule, which corresponds to $\geq 97\%$ labeling.

The synthesis of 1b is outlined in Scheme II. The preparation of 3,6,9-trioxaundecane-1,11-diol-1,1,11,11- d_4 di-p-toluenesulfonate (3b) has been described previously (5), and its reaction with 2,6-bis(hydroxymethyl)pyridine (2b) in THF-water containing potassium t-butoxide yielded 1b. By mass spectrometry 1b contained \geq 3.96 atoms of excess deuterium per molecule, which corresponds to \geq 99% labeling.

The preparation of 1c is outlined in Scheme III. The disodium salt of diol 2b on reaction with methyl bromoacetate in THF yielded 2,6-bis[(methoxycarbonylmethoxy)methyl]pyridine (4), which was reduced with sodium borodeuteride in ethanol-Od to give 2,6-bis[(2-hydroxyethoxy-2,2,2',2'-d₄)methyl]pyridine (5a). Reaction of the disodium salt of diol 5a with 3-oxapentane-1,5-diol di-p-toluenesulfonate (6a) yielded 1c, which contained

Scheme IV

I.
$$(CH_3)_3COK/$$

THF- H_2O

2. 3c

 $C = (p-TosOCH_2CH_2OCD_2CH_2)_2O$

Id

≥ 3.96 atoms of excess deuterium per molecule.

The synthesis of 1d is outlined in Scheme IV. reaction of previously described (5) 3,6,9-trioxaundecane-1,11-diol-4,4,8,8- d_4 (3c) with 2b in THF-water containing potassium t-butoxide gave 1d, which by mass spectrometry contained ≥ 3.96 atoms of excess deuterium per molecule. The preparation of 1e is summarized in Scheme Exchange of the methylene protons of dimethyl 2,2'-oxydiacetate with methanol-O-d-sodium methoxide followed by reduction of the resultant labeled diester with lithium aluminum hydride gave 3-oxapentane-1,5diol-2,2,4,4-d₄, which was converted to 3-oxapentane-1,5-diol-2,2,4,4- d_4 di-p-toluenesulfonate (6b). Reduction of 4 with sodium borohydride in ethanol yielded 2,6-bis-[(2-hydroxyethoxy)methyl]pyridine (5b), and treatment of its disodium salt with 6b gave 1e, which by mass spectrometry contained ≥ 3.96 atoms of excess deuterium per molecule.

EXPERIMENTAL

The ¹H nmr spectra were obtained with a Varian HA-100 spectrometer, and deuteriochloroform was used as solvent with tetramethylsilane as internal standard. Mass spectra were recorded with direct sample insertion on a Varian MAT CH-5 mass spectrometer. The ionizing voltage was 70 eV, the filament current 300 µA, and the source temperature 150°. For analytical and preparative gas-liquid chromatography (glc), a 6 ft. x 1/4 in. aluminum column packed with 1% SE-30 on 60-80 mesh AW-DMCS Chromosorb W was used with helium as carrier gas. For column chromatography, unless specified otherwise, neutral alumina was used with 1% ethanol in dichloromethane as eluent. Tetrahydrofuran (THF) was purified by distillation from lithium aluminum hydride under nitrogen. Microanalyses were performed by Huffman Laboratories, Wheat Ridge, Colorado, and by Galbraith Laboratories, Knoxville, Tennessee. All melting and boiling points are uncorrected.

2,6-Bis(hydroxymethyl)pyridine (2b).

The preparation of this material followed the procedure of Cram and co-workers (6). From 15.5 g. (9.28 mmoles) of 2,6-pyridinedicarboxylic acid (Aldrich), 16.7 g. (92%) of dimethyl 2,6-pyridinedicarboxylate, m.p. 119-121° [lit. (7) m.p. 121°], was obtained. Then, 14.5 g. (7.44 mmoles) of the diester yielded 10.0 g. (97%) of crude **2b**, which was recrystallized from chloro-

form to give purified material, m.p. $112\text{-}114^\circ$ [lit. (8) m.p. $114.5\text{-}115^\circ$].

3,6,9,12,15-Penta oxa-21-azabicyclo[15.3.1]heneicosa-1(21),17,19-triene (pyridyl-18-crown-6, 1)(4).

A solution of 0.347 g. (2.50 mmoles) of **2b** in 15 ml. of THF was added to a stirred solution of 0.600 g. (5.36 mmoles) of potassium t-butoxide in 10 ml. of THF under nitrogen. Then a solution of 0.650 g. (1.29 mmoles) of 3,6,9-trioxaundecane-1,11diol di-p-toluenesulfonate (3a) (9) in 10 ml. of THF was added, and the mixture was refluxed for 30 minutes. After 0.5 ml. of water was added, the reaction mixture was refluxed under nitrogen for 15 hours, cooled to 25°, filtered, and concentrated on a rotary evaporator. The resulting oil was column-chromatographed to give 0.40 g. (54%) of crude 1, which solidified at 0°. This material was sublimed at 80° (0.05 mm) to give 1, m.p. 39.0-¹H nmr: δ 3.52 (s, 8H, 39.5° [lit. (4) m.p. 40-41°]; CH₂CH₂OCH₂CH₂ nonadjacent to pyridine ring), 3.64 (A₂B₂, 8H, $ArCH_2OCH_2CH_2$), 4.68 (s, 4H, $ArCH_2$), 7.16 (d, J = 8 Hz, 2H, 3-ArH), 7.58 (t, J = 8 Hz, 1H, 4-ArH).

3,6,9,12,15-Pentaoxa-21-azabicy clo[15.3.1]heneicosa-1(21),17,19-triene-2,2,16,16- d_4 (1a).

With the procedure for 2b, 0.810 g. (4.16 mmoles) of dimethyl 2.6-pyridinedicarboxylate was converted to 0.230 g. (40%) of 2.6-bis(hydroxymethyl-d2)pyridine (2a) with sodium borodeuteride (99% D) and ethanol-O-d (99% D). To a stirred slurry of 0.067 g. (2.8 mmoles) of sodium hydride in 50 ml. of THF under nitrogen was added a solution of 0.200 g. (1.40 mmoles) of 2a in 50 ml. of THF. After gas evolution ceased, a solution of 0.735 g. (1.46 mmoles) of 3a (9) in 50 ml. of THF was added, and the reaction mixture was refluxed for 16 hours, cooled to 25°, filtered, and concentrated on a rotary evaporator. The residue was columnchromatographed to give 0.090 g. (21%) of 1a, m.p. 39-39.5° [lit. (4) m.p. 40-41° for 1]; ¹H nmr: δ 3.52 (s, 8H, CH₂CH₂-OCH₂CH₂ nonadjacent to pyridine ring), 3.64 (A₂B₂, 8H, ArCD₂OCH₂CH₂), 4.68 (bs, < 0.1 H, residual ArCHD), 7.16 (d, J = 8 Hz, 2H, 3-ArH), 7.58 (t, J = 8 Hz, 1H, 4-ArH). A sample was purified by preparative glc (200°), and by mass spectrometry it contained 3.88 atoms of excess deuterium per molecule.

In a separate purification, ethanol-O-d was used along with alumina that had been dried for 48 hours at 400° and treated with deuterium oxide. This modification did not result in 1a with a higher deuterium content.

3,6,9,12,15-Pentaoxa-21-azabicyclo[15.3.1]heneicosa-1(21),17,19-triene-4,4,14,14 - d_4 (**1b**).

With the procedure used for 1, 1.31 g. (2.59 mmoles) of 3,6,9-trioxaundecane-1,11-diol-1,1,11,11-d₄ di-*p*-toluenesulfonate (3b) (5) and 0.348 g. (2.49 mmoles) of 2b gave 0.260 g. (33%) of 1b, m.p. 39-39.5° [lit. (4) m.p. 40-41° for 1]; 1 H nmr: δ 3.52 (s, 8H, CH₂CH₂OCH₂CH₂), 3.64 (s, 4H, CH₂CD₂), 4.68 (s, 4H, ArCH₂), 7.16 (d, J = 8 Hz, 2H, 3-ArH), 7.58 (d, J = 8 Hz, 1H, 4-ArH). A sample was purified by preparative glc (200°), and by mass spectrometry it contained \geq 3.96 atoms of excess deuterium per molecule.

3,6,9,12,15-Pentaoxa-21-azabicyclo [15.3.1]heneicosa-1(21),17,19-triene-5,5,13,13- d_4 (1c).

A procedure of Cram and Newcomb (10) for 1 with appropriate deuterated reagents gave 1c. A solution of 5.80 g. (42.0 mmoles) of 2b in 200 ml. of THF was added to a stirred slurry of 4.90 g. (0.204 mole) of sodium hydride in 200 ml. of THF under nitrogen. After gas evolution ceased, 40.0 g. (0.260 mole) of

methyl bromoacetate was added, and the mixture was refluxed for 8 hours, cooled to 25° , filtered, and concentrated on a rotary evaporator. The residue was dissolved in a mixture of 100 ml. of water and 250 ml. of dichloromethane, and the resulting aqueous layer was acidified to pH 5 with 5% hydrochloric acid. The dichloromethane layer was dried over sodium sulfate, and rotary evaporation left crude product which was column-chromatographed on silica gel with acetone-dichloromethane elution to give 3.86 g. (34%) of 2,6-bis[(methoxycarbonylmethoxy)methyl]-pyridine (4) (10) as an oil; ^{1}H nmr: δ 3.69 (s, 6H, CH₃), 4.16 (s, 4H, OCH₂CO), 4.67 (s, 4H, ArCH₂), 7.31-7.79 (m, 3H, ArH).

A solution of 1.20 g. (4.24 mmoles) of 4 and 0.540 g. (12.9 mmoles) of sodium borodeuteride (99% D) in 50 ml. of ethanol-O-d (99.9% D) was refluxed under nitrogen for 10 hours. The ethanol-O-d was removed by distillation, and the residue was dissolved in 30 ml. of water and extracted four times with dichloromethane. The combined extracts were dried over sodium sulfate and rotary evaporated to leave 0.58 g. (58%) of 2,6-bis-[(2-hydroxyethoxy-2,2,2',2'-d₄)-methyl]pyridine (5a) (10) as an oil; ¹H nmr: δ 3.63 (s, 4H, CH₂CD₂), 4.51 (bs, 2H, OH), 4.59 (s, 4H, ArCH₂), 7.12-7.71 (m, 3H, ArH).

A solution of 0.470 g. (2.00 mmoles) of **5a** in 20 ml. of THF was added to a stirred slurry of 0.125 g. (5.23 mmoles) of sodium hydride in 20 ml. of THF under nitrogen. After gas evolution ceased, a solution of 0.828 g. (2.00 mmoles) of 3-oxapentane-1,5-diol di-p-toluenesulfonate (**6a**) (11) in 20 ml. of THF was added, and the mixture was refluxed for 16 hours, cooled to 25°, filtered, and concentrated on a rotary evaporator. Column-chromatography of the residue gave 0.262 g. (44%) of **1c**, m.p. 39-39.5° [lit. (4) m.p. 40-41° for **1**]; 1 H nmr: 8 3.52 (s, 8H, CH₂CH₂OCH₂CH₂), 3.64 (s, 4H, CH₂CD₂), 4.69 (s, 4H, ArCH₂), 7.16 (d, J = 8 Hz, 3H, 3-ArH), 7.58 (t, J = 8 Hz, 1H, 4-ArH). A sample was purified by preparative glc (200°), and by mass spectrometry it contained \geq 3.96 atoms of excess deuterium per molecule.

3,6,9,12,15-Pentaoxa-21-azabicy clo[15.3.1]heneicosa-1(21),17,19-triene-7,7,11,11- d_4 (**1d**).

With the procedure used for 1, 0.348 g. (2.49 mmoles) of **2b** and 1.31 g. (2.59 mmoles) of 3,6,9-trioxaundecane-1,11-diol-4,4,8,8- d_4 (**3c**) (5) gave 0.220 g. (28%) of **1d**, m.p. 39-39.5° [lit. (4) m.p. 40-41° for 1]; ¹H nmr: δ 3.52 (s, 4H, CD₂CH₂-OCH₂CD₂), 3.64 (A₂B₂, 8H, CH₂CH₂), 4.68 (s, 4H, ArCH₂), 7.16 (d, J = 8 Hz, 2H, 3-ArH), 7.58 (t, J = 8 Hz, 1H, 4-ArH). A sample was purified by preparative glc (200°), and by mass spectrometry it contained \geq 3.96 atoms of excess deuterium per molecule.

3,6,9,12,15-Pentaoxa-21-azabicyclo[15.3.1]heneicosa-1(21),17,19-triene-8,8,10,10- d_4 (**1e**).

In 8.0 g. (0.24 mole) of methanol-O-d (99% D) under nitrogen 5 mg. (0.2 mmole) of sodium was dissolved, and then 1.00 g. (8.95 mmoles) of dimethyl 2,2'-oxydiacetate (5) was added. The mixture was stirred at 25° for 24 hours, and methanol-O-d was removed by vacuum distillation. Without the use of additional

sodium, the above procedure was repeated eight times with fresh 8.0 g. portions of methanol-O-d. After the last exchange, 8 drops of 20% deuterium chloride in deuterium oxide (99% D, Aldrich) were added, and methanol-O-d was distilled. resulting diester was reduced with lithium aluminum hydride in ether in standard fashion to give 0.22 g. (33%) of 3-oxapentane-1,5-diol-2,2,4,4-d₄, which was converted (9) to the corresponding di-p-toluenesulfonate 6b. Reduction of diester 4 with sodium borohydride in ethanol with the procedure used for preparation of 5a gave 2.6-bis[(2-hydroxyethoxy)methyl] pyridine (5b). Then with the procedure employed for 1c, 0.282 g. (1.22 mmoles) of 5b and 0.440 g. (1.05 mmoles) of 6b yielded 0.0820 g. (22%) of **1e.** m.p. 39-39.5° [lit. (4) m.p. 40-41° for 1]; ¹H nmr: δ 3.52 (s, 4H, CH₂CD₂OCD₂CH₂), 3.64 (A₂B₂, 8H, CH₂CH₂), 4.68 (s, 4H, $ArCH_2$), 7.16 (d, J = 8 Hz, 2H, 3-ArH), 7.58 (t, J = 8 Hz, 1H, 4-ArH). A sample was purified by preparative glc (200°), and by mass spectrometry it contained ≥ 3.96 atoms of excess deuterium per molecule.

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